

THEORETICAL STUDY OF ELECTRONIC PROPERTIES OF Ag_2Te

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The electronic band structure and density of state calculations were performed for the low-temperature modification of a silver chalcogenide - $\beta\text{-Ag}_2\text{Te}$ through Atomistic Simulation Software Quantum ATK. The structures are characterized by three, four and five coordinations of silver by the chalcogen. According to the band structure calculations, $\beta\text{-Ag}_2\text{Te}$ is semiconductor with an about 0.1–0.2 eV forbidden zone. The calculations have shown that $\beta\text{-Ag}_2\text{Te}$ has a very low DOS in the energy range from about –0.1 to +0.5 eV.

Keywords: electronic band structure, chalcogenide, density of state, silver, semiconductor.

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INTRODUCTION

Ag_2Te , one of silver chalcogenides, is known as Hessite mineral in nature. It was used as ionic conductor at high temperature phase. The zone structure of Ag_2Te was studied through local spherical wave (LSW) method. In accordance with the LSW calculation of the zone structure, this compound is semi-metallic having approximately $0.1 \div 0.2 \text{ eV}$ energy cover (forbidden zone). It undergoes a phase transition below 417K into the phase, a narrow gap semiconductor, where the ion migration is frozen and the compound is nonmagnetic. Ag_2Te changes from a regular structure to a chaotic structure at low temperatures. The transition temperature of Ag_2Te is 1450°C . The gap of $\beta\text{-Ag}_2\text{Te}$ is in the range of several tens *meV*, the mobility of carriers is high and the effective mass is of the order of $102 m_0$ (m_0 is the free electron mass) [1, 2].

It is known that the low-temperature (β) phases for Ag_2Te has large and positive magneto-resistance. This compound is non-magnetic. In order to understand the origin of magneto-resistance, it is important to have profound knowledge about its electronic structure.

The Ag chalcogenides are of great importance first of all due to the high ionic conductivity (of Ag^+) of the high temperature (α) structure and electronic conductivity. The high ionic conductivity in the α phase of the Ag_2Te is due to the static distribution of silver atoms in the lattice. These compounds show a transition at rather low temperatures from the β structure to α structure. The transition temperature is 139°C for Ag_2Se and 145°C for Ag_2Te [3].

The electrical transport properties of Ag_2Te have been extensively studied and reported [4-7]. In these studies, a very small temperature range is studied and the phase transition environment is studied. According to the results of the band structure calculations it is small gap (20-50 *meV*) semiconductor. The deviation from stoichiometry determines both the electronic conductivity and activation energy. The mobility of electrons and holes is high with small effective mass. Concentration of electron - carriers is about $3 \times 10^{18} \text{ cm}^{-3}$. Earlier, the lattice parameters of $\beta\text{-Ag}_2\text{Te}$

were calculated by Van der Lee, A. and J.L. de Boer. Lattice parameters and coordinates of $\beta\text{-Ag}_2\text{Te}$ are from an accurate single crystal refinement by the mentioned researchers [7].

RESULTS

In this article the electronic band structure and density of state were calculated for $\beta\text{-Ag}_2\text{Te}$ using the Atomistic Simulation Software Quantum ATK method [8].

Figure 1 shows the atomic structure of $\beta\text{-Ag}_2\text{Te}$. As it can be seen from the figure, the extraordinary superconducting properties of $\beta\text{-Ag}_2\text{Te}$ is associated with these linear chains of Ag atoms.

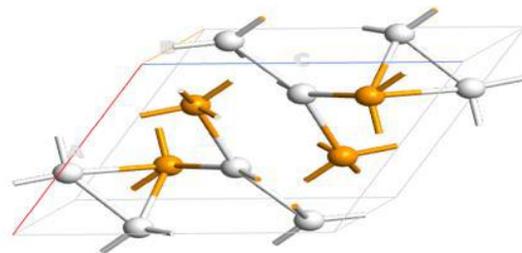


Fig. 1. Atomic structure of $\beta\text{-Ag}_2\text{Te}$.

Besides the metal–chalcogen distances corresponding to bonding, there are also metal-metal-distances as in pure metals. Table 1 presents the interatomic distances for $\beta\text{-Ag}_2\text{Te}$. $\beta\text{-Ag}_2\text{Te}$ has two types of silver atoms in its structure. Both types of Ag atoms have a four-fold coordination by Te atoms. The Ag-Ag distances range from 2.84 to 3.13 Å. The chalcogen packing is distorted face-centered cubic.

The band structure of $\beta\text{-Ag}_2\text{Te}$ plotted for high symmetry points $\Gamma\text{-X-M}\Gamma\text{-R-X}$ at $T=300\text{K}$ is given in figure 2. The band structure calculated based on density functional theory within spin-polarized PBE-GGA approximation.

Figure 3 presents some calculated results of the density of state (DOS) constructed for *s*-, *p*- and *d*-electrons of $\beta\text{-Ag}_2\text{Te}$.

Interatomic distances between atoms in β -Ag₂Te crystal (\AA)

Ag(1)-Te	2.877	2.895	2.965	3.016
Ag(2)-Te	2.842	2.905	3.011	3.034
Ag(1)- Ag(1)	2.841	3.010	-	-
Ag(2)- Ag(2)	3.053	-	-	-
Ag(1)- Ag(2)	3.061	3.133	2.909	

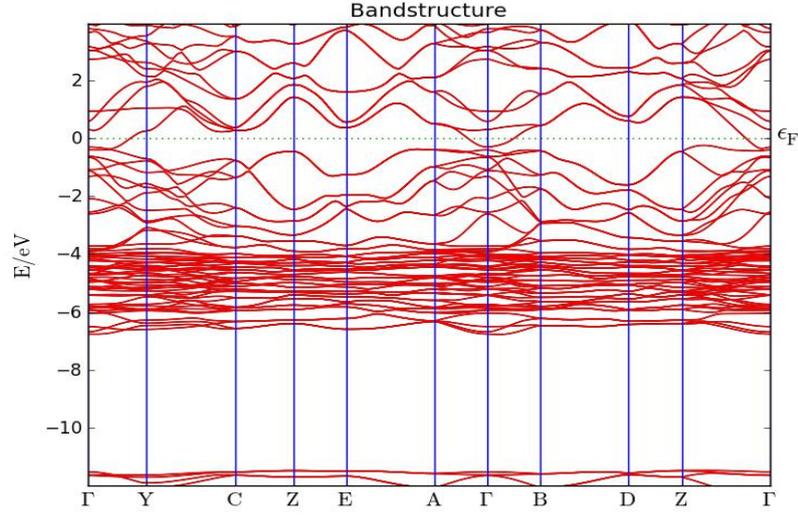


Fig. 2. Electronic band structure of β -Ag₂Te.

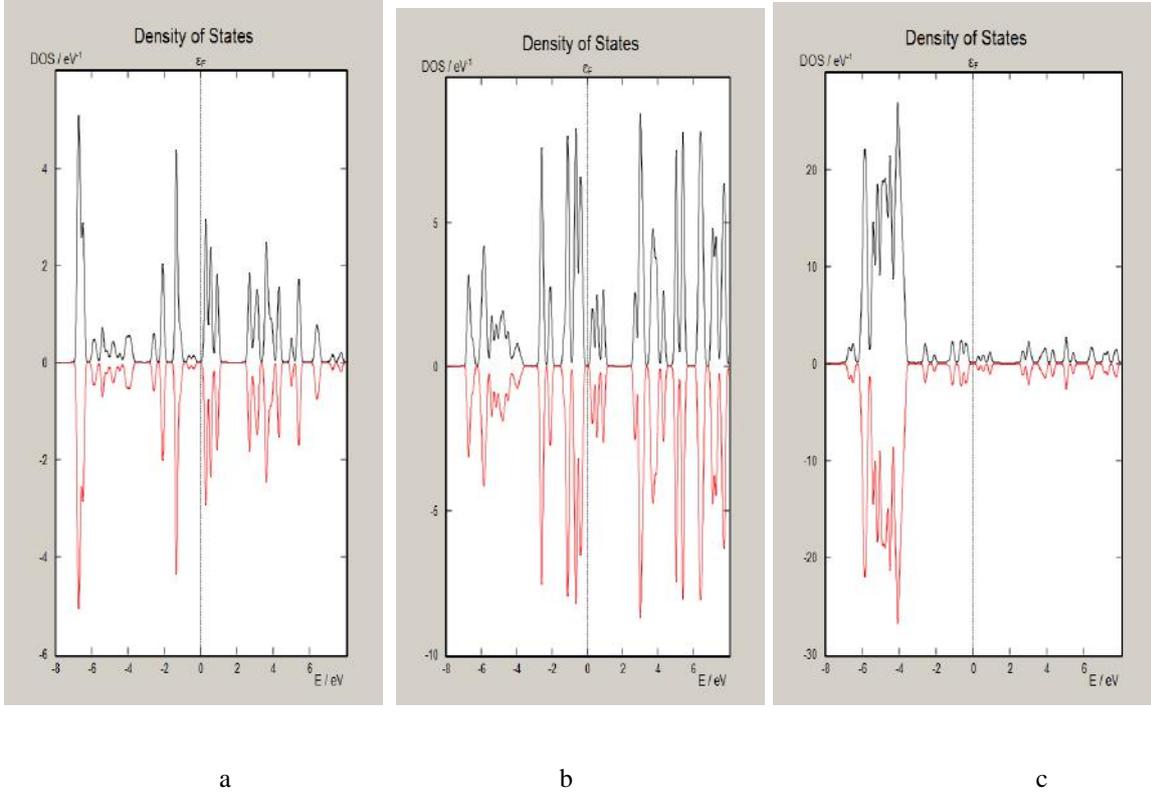


Fig. 3. DOS for s - (a); p - (b) and d - (c) electrons of β -Ag₂Te.

CONCLUSION

The electronic band structure and density of state were calculated for low temperature β -Ag₂Te using the Atomistic Simulation Software Quantum ATK method. It was found that the extraordinary superconducting properties of β -Ag₂Te is associated with the linear

chains of Ag atoms. There is a small overlap between the conduction and valence band of β -Ag₂Te. In addition, DOS was calculated for s - (a); p - (b) and d - (c) electrons of β -Ag₂Te.

Based on the calculations, we can say that β -Ag₂Te have a very low DOS in the energy range from about -0.1 to $+0.5$ eV.

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